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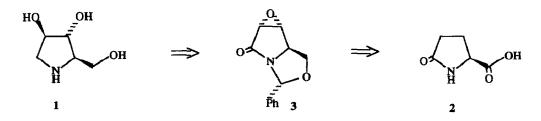
A Short Diastereoselective Synthesis of the Natural (2R, 3R, 4R)-2-Hydroxymethyl-3,4-Dihydroxypyrrolidine

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Abstract: 1,4-Dideoxy-1,4-imino-D-arabinitol 1, a glycosidase inhibitor constituent of <u>Arachniodes standishii</u>¹ and <u>Angylocalyx boutiqueanus</u>² was synthesized from (S)-pyroglutamic acid through regioselective ring opening of the epoxide 3.

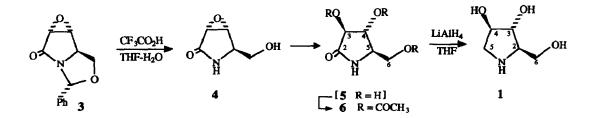
Naturally occuring polyhydroxylated pyrrolidines have received much attention due to their ability to inhibit glycosidases.³⁻⁵ Among them, 1,4-dideoxy-1,4-imino-D-arabinitol 1, isolated from Arachniodes standishii ¹ and Angylocalyx boutiqueanus ², is known as a potent inhibitor of yeast α -glucosidase, as a potential AIDS virus replication inhibitor⁶ and exhibits several other biological activities.^{7a}



Therefore, much efforts have been directed to synthesize the compound 1, structurally related to sugar, from carbohydrates.^{3,7} Some other routes developed to achieve the same goal involved the use of aldolases to catalyze aldol condensations, prior to reductive amination and cyclization to the pyrrolidine ring.⁸ In addition, a synthesis of 1,4-dideoxy-1,4-imino-D-arabinitol 1 from (S)-pyroglutamic acid 2 has already been described.⁹ As part of a program to synthesize several bioactive molecules using (S)-pyroglutamic acid 2 as a chiral precursor¹⁰, a straightforward diastereoselective synthesis of 1 from the epoxy pyrrolidone $3^{10b,11}$ is described here.

The high regioselectivity that we observed in the oxirane ring opening of 3 by hydride ion^{10b}, led us to check its regioselective opening by other nucleophiles, as well. Thus, the epoxy lactam 3 could be a valuable intermediate in the synthesis of several natural products, particularly the aqueous acidic hydrolysis of 3 could lead to a direct precursor of (2R, 3R, 4R)-2-hydroxymethyl-3,4-dihydroxypyrrolidine 1.

This step could be achieved in one-pot together with the deprotection of both primary alcohol and nitrogen atom of 3 (CF₃CO₂H-THF-H₂O, 80°C) to afford the trihydroxylated lactam 5, through the 3,4-epoxy-5-hydroxymethyl-pyrrolidin-2-one 4.¹² The water soluble crude polyhydroxylated pyrrolidone 5 was directly converted (excess Ac₂O, pyridine, r.t., 24 h.) to its triacetate 6^{13} , isolated chromatographically in 43% yield from 3.



The configurations of 6 were assigned by ¹H NMR. The doublet at 5.46 ppm (1H, $J_{3,4} = 6$ Hz) was attributed to the proton H-3, whereas the proton H-4 gives rise to a doublet of doublet at 5.28 ppm ($J_{3,4} = J_{4,5} =$ 6 Hz). The NOE observed between H-3 and H-5 and between H-4 and H-6 support the indicated configurations. The lactam 6 was reduced by an excess of LiAlH4 in refluxing THF (4h.) to afford 1,4dideoxy-1,4-imino-D-arabinitol 1 (74%).¹⁴ This simple scheme could be applied to the commercially available (R)-pyroglutamic acid to obtain the enantiomer 1,4-dideoxy-1,4-imino-L-arabinitol, which has been reported to be a potent inhibitor of the cytopathic effect of AIDS virus at non-cytotoxic concentrations.⁶

The use of the epoxy pyrrolidone 3 in further syntheses is under current investigation in our laboratory.

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 A very recent synthesis of (2S,3S)-3-hydroxyproline from 3 prompted us to disclose our own results :
- Herdeis, C.; Hubmann, H.P.; Lotter, H. Tetrahedron Asymmetry 1994, 5, 119-128.
- 12. Previously characterized as its acetate, see reference 10b.
- 13. 6: mp : 79-81°C; $[\alpha]_D^{30} = +45$ (c = 0.78, CHCl₃); IR : 1742 cm⁻¹; ¹H NMR [250 MHz, CDCl₃, d = 0 : TMS, J (Hz)] : 6.32 (bs, 1H, NH), 5.46 (d, 1H, J = 6, H-3), 5.28 (dd, 1H, J = J' = 6, H-4), 4.48 (dd, 1H, J = 12, J' = 3.5, Ha-6), 4.04 (dd, 1H, J = 12, J' = 6, Hb-6), 3.78 (m, 1H, H-5), 2.17 (s, 3H, CH₃), 2.12 (s, 3 CH₃), 2.11 (s, 3H, CH₃); ¹³C NMR (75 MHz, CDCl₃): 170.81 (CO), 170.24 (CO), 170.00 (CO), 169.46 (CO), 74.10 (CHO), 74.07 (CHO), 63.75 (CH₂O), 55.62 (CHN), 20.71 (CH₃); MS (m/z) : 274 (M^{+,} + H), 213, 200, 171 (100%), 140, 98 ; Anal. calcd for $C_{11}H_{15}NO_7$: C, 48.35 ; H, 5.53 ; N, 5.13. Found : C, 48.36; H, 5.28; N, 5.17.
- 14. 1 (hydrochloride) $[\alpha]_D^{33} = +30$ (c = 0.41, H₂O); comparison of ¹H NMR (300 MHz), ¹³C NMR (75 MHz, D₂O), and CIMS data.³

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